

Using FTIR-Spectroscopy to Measure Atmospheric Trace Gas Concentrations Over Toronto

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Mexico



U of Toronto Atmospheric Observatory (TAO)

Location:
43.66N, 79.40W,
174 masl

↓ Solar tracker



- Primary instrument:
 - ABB Bomem DA8 Fourier transform infrared (FTIR) spectrometer
- Measurements started in 2002
- Urban site (downtown Toronto)

Scientific Objectives

- To assess biomass burning emissions and quantify its emission factors
- To investigate the daily, seasonal, and interannual trace gas variability and trends in an urban setting (downtown Toronto)
- To characterize the origin of urban pollution events (local or long-range transport) with models (e.g., the GEOS-Chem chemical transport model)

Biomass Burning

- Biomass burning emissions can negatively affect air quality
- Emissions can have photochemical and radiative forcing effects, particularly when plumes are transported to Arctic regions (Amiro et al., 2001)
- Quantifying biomass burning emissions and understanding their transport poses a challenge
 - Types of vegetation burned, the combustion phase (smoldering vs. flaming), and atmospheric conditions at the time of the fire must all be accounted for when quantifying emissions
 - Emitted gases may also undergo chemistry in the atmosphere during transport, which must be accounted for

Biomass Burning Enhancement Events

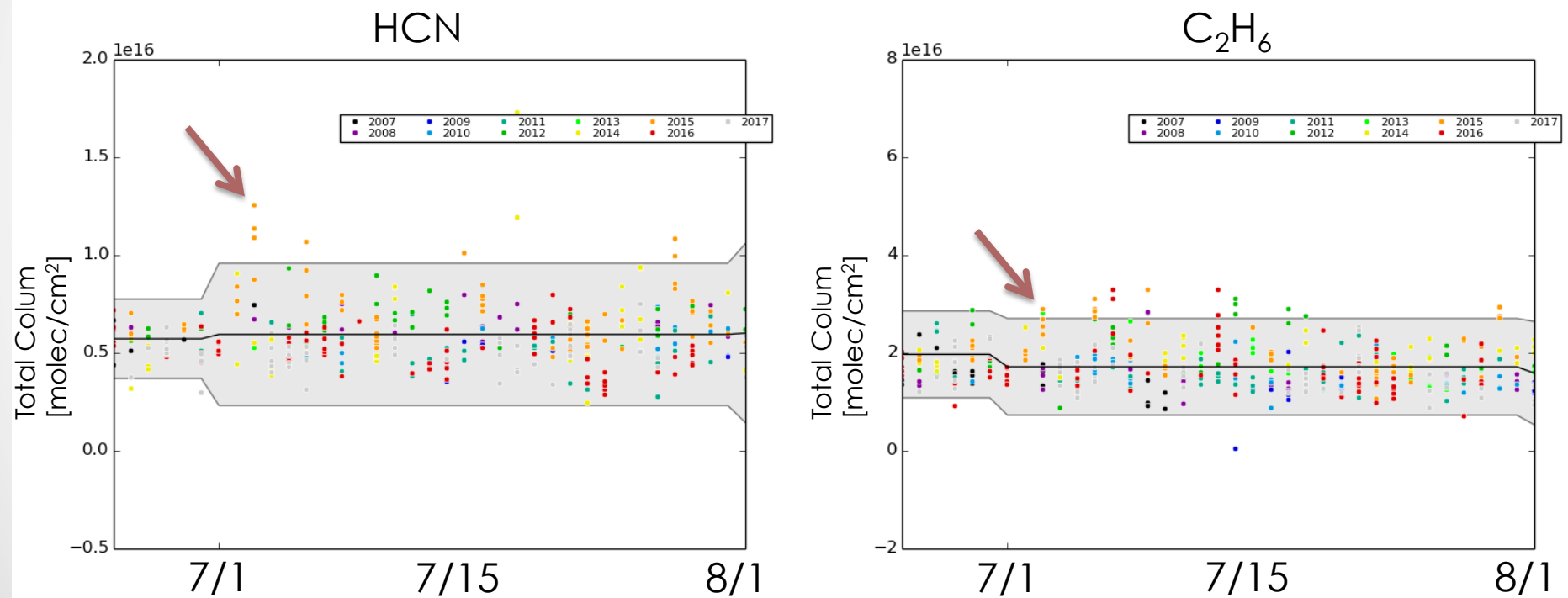
- Wildfires emit numerous chemical species including CO, C₂H₆, HCN, HCOOH, H₂CO, C₂H₂ and others
- Given their long lifetimes, CO, HCN and C₂H₆ are good tracers of biomass burning events (e.g., Viatte et al., 2013)
- Formic acid and formaldehyde have shorter lifetimes
 - Typical lifetimes (τ) are: 61 days (CO), 45 days (C₂H₆), 150 days (HCN), \sim 4 days (HCOOH) and \sim 1 day (H₂CO) (Viatte et al., 2013, Millet et al., 2015, Pommier et al., 2017)

Biomass Burning Analysis

- Analyzing the 2015 early July biomass burning enhancement event observed at TAO
 - Simultaneous enhancements of HCN, CO, C₂H₆ and HCOOH
 - HCOOH is an important gas in the atmosphere as it greatly contributes to free precipitation acidity
- FLEXPART used for source attribution
- Travel-time estimated with HYSPLIT
 - Travel-time used to estimate decay in the atmosphere
- Emission factors and emission ratios reported

Early July 2015

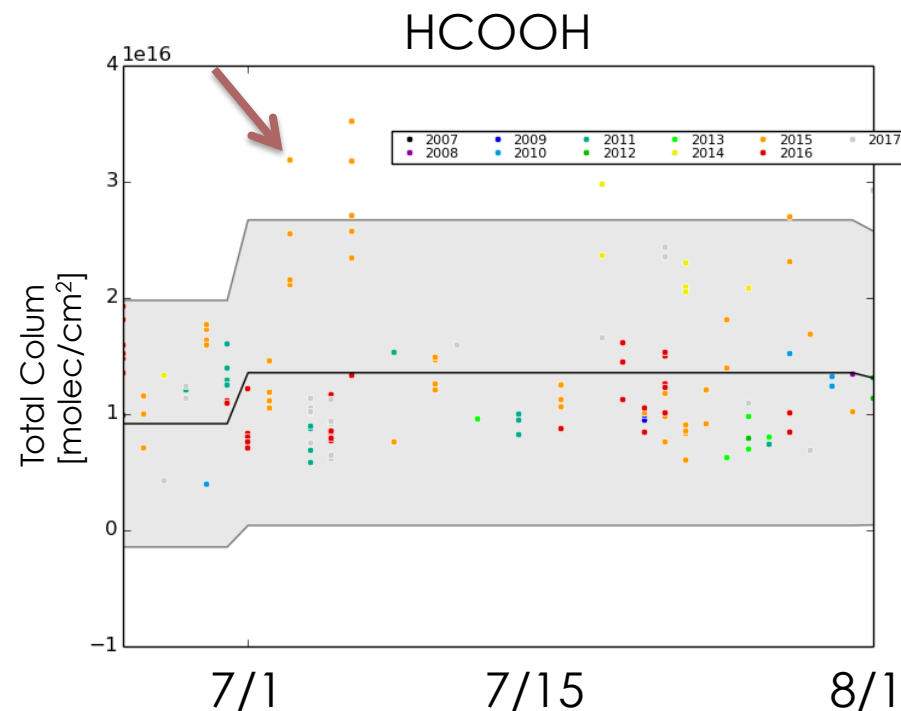
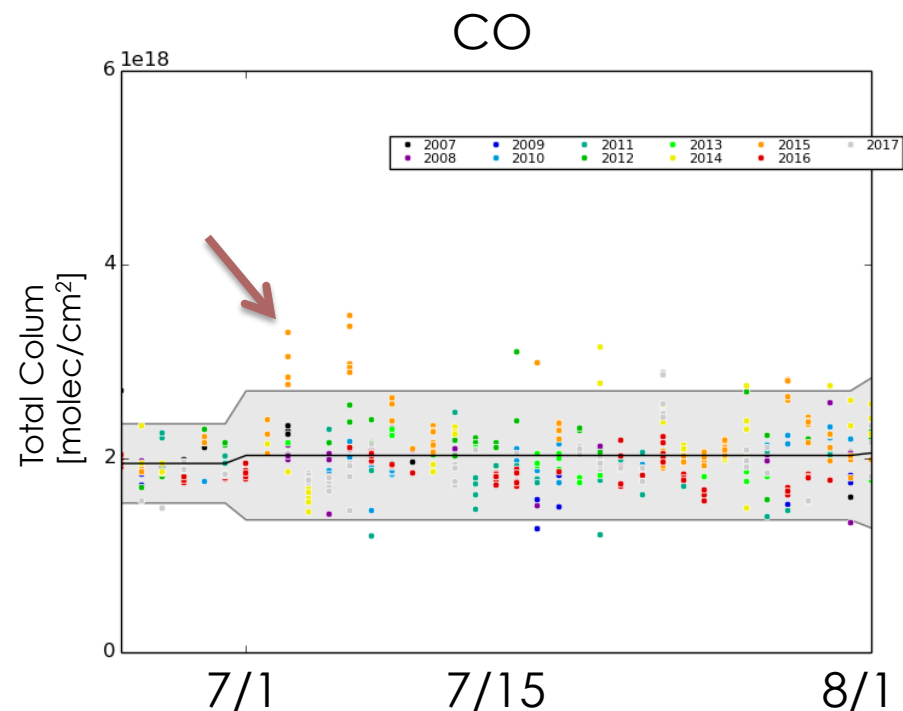
- Notice the sharp peak in HCN on July 3–6
- C_2H_6 enhancement is less obvious, likely due to C_2H_6 having local sources



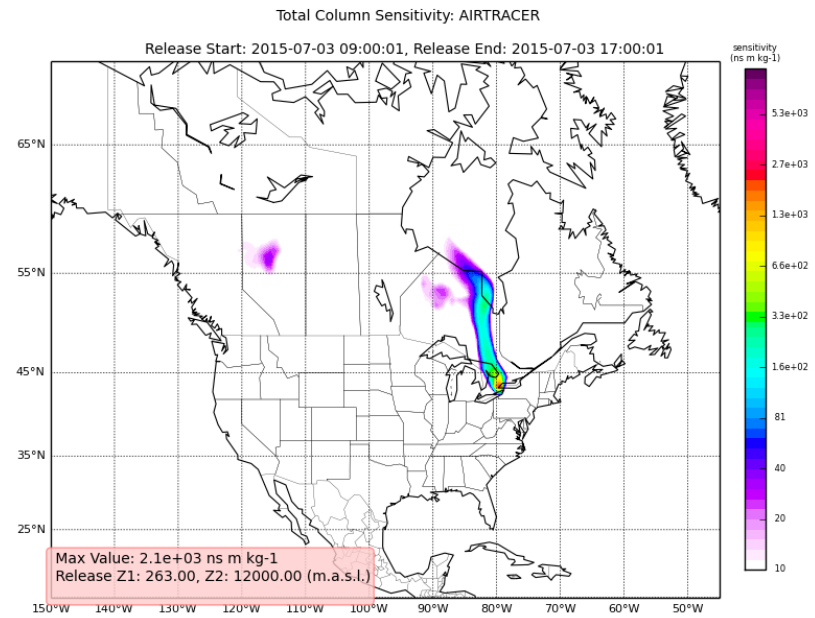
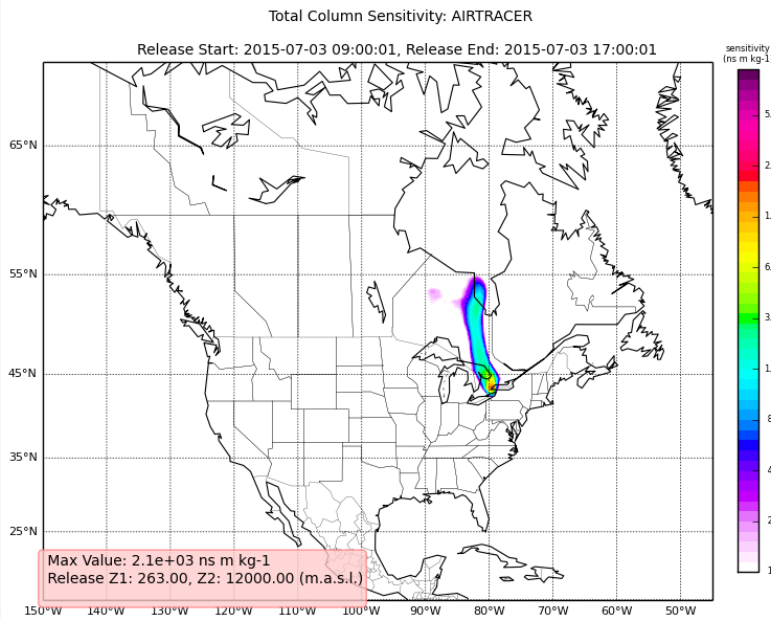
- Orange dots are 2015 data points
- Gray band indicates 2 standard deviations above and below the monthly mean

Early July 2015 Cont.

- CO is also clearly enhanced on July 3rd and 6th
- HCOOH is concurrently enhanced

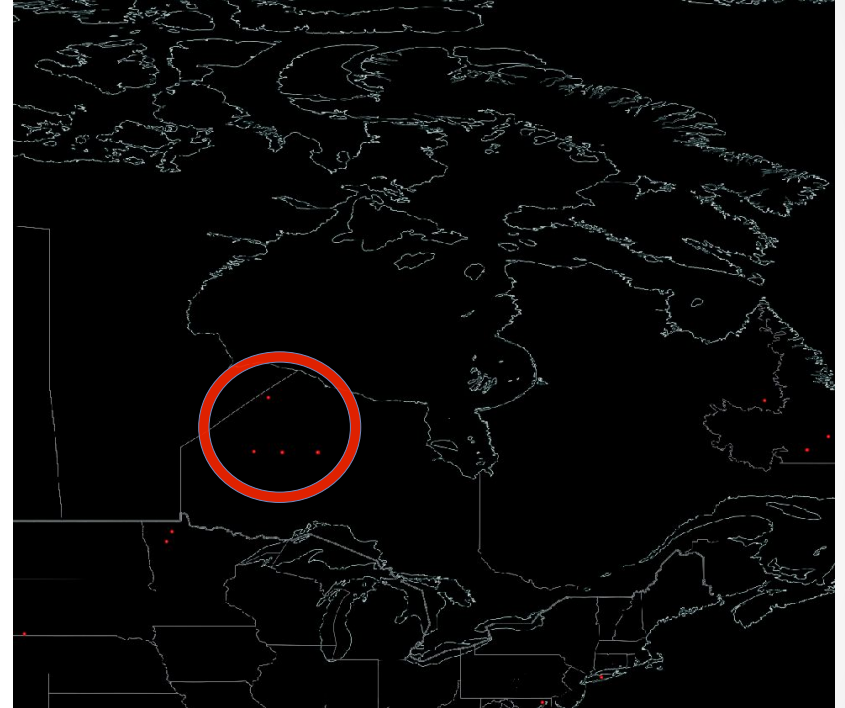
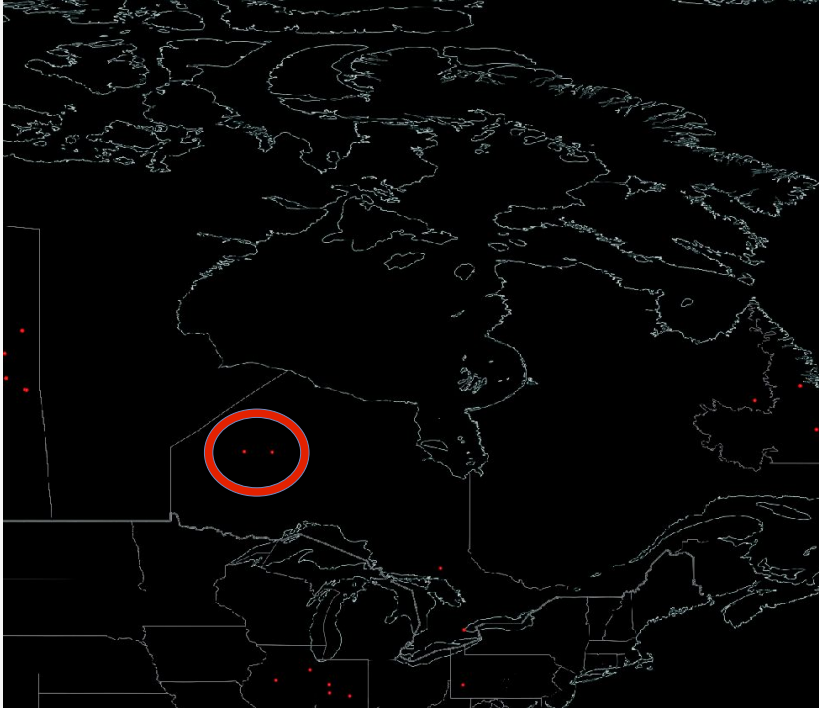


July 3rd Analysis with FLEXPART



FLEXPART 2-day and 3-day
back-trajectory runs (both July
3rd release), left and right
respectively

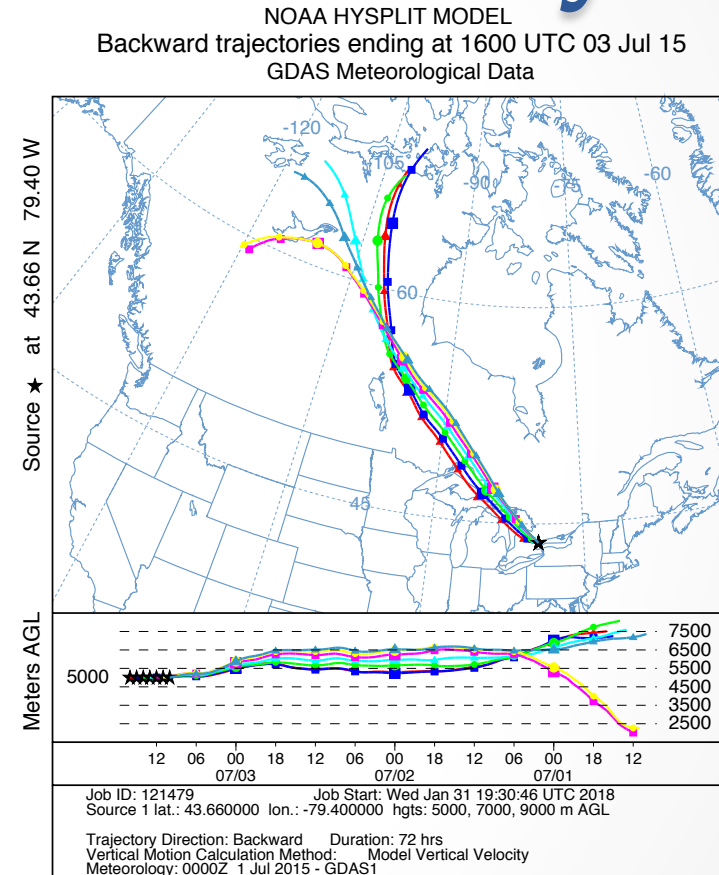
Comparison with MODIS Fire Data



- MODIS Fire Product is an online tool for detecting fire anomalies
 - MODIS is onboard satellites TERRA and AQUA
- Fire events detected by MODIS on July 1st and 2nd, left and right respectively

Travel Time and Decay

- Chemical species decay exponentially in the atmosphere
 - i.e., $[x]_t = [x]_0 e^{-t/\tau}$
 - Where τ denotes the lifetime
- HYSPLIT back trajectory was run with GDAS meteorological field for travel time estimation
- For this event travel time was taken to be 2 days



Interpreting the Data

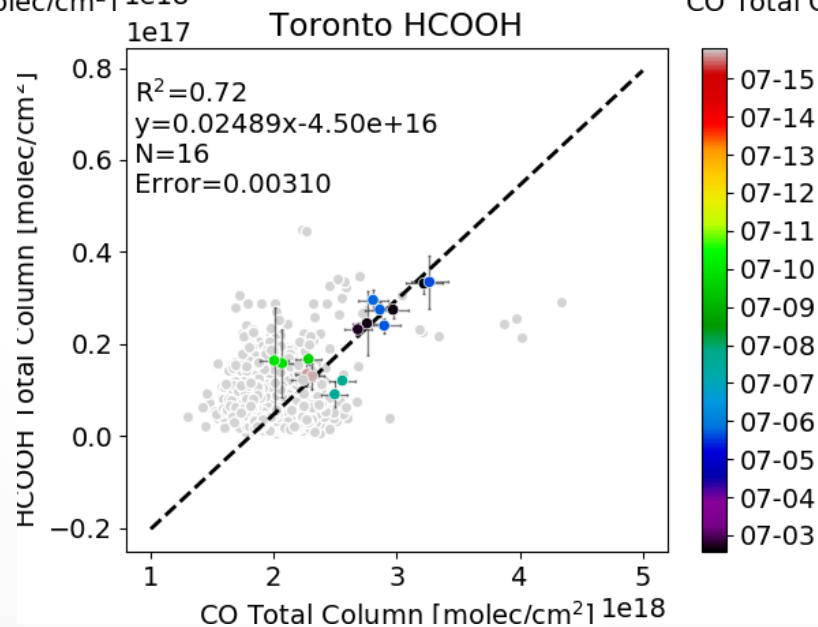
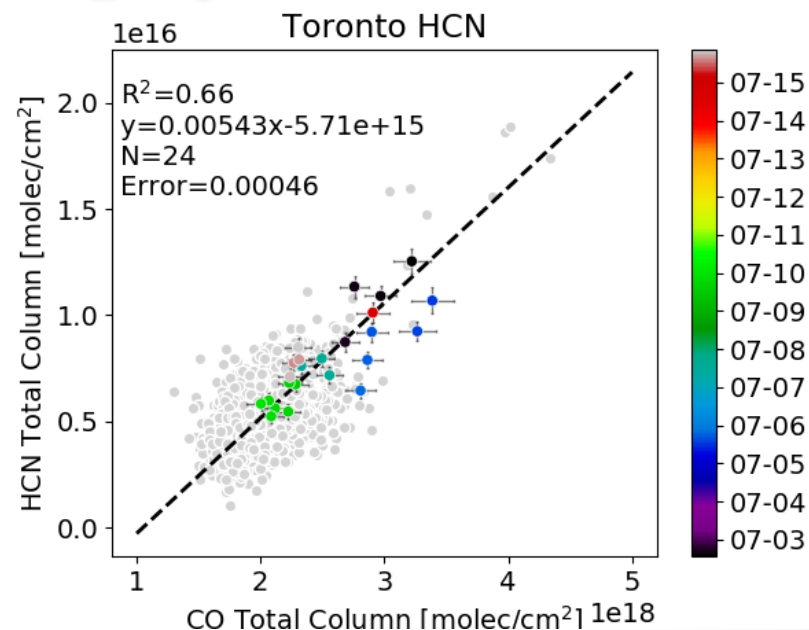
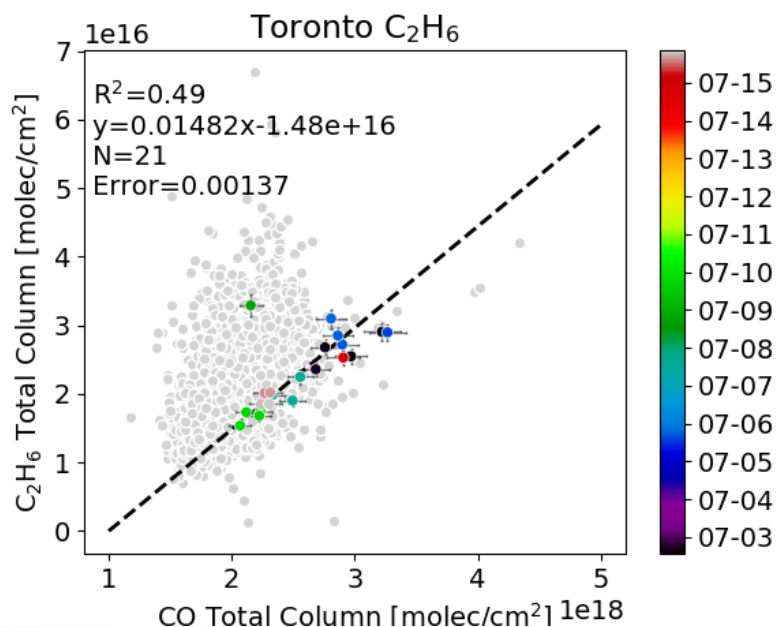
- Enhancement Ratio (EnhR)
 - Ratio of column of species of interest to column of CO
 - Calculated for HCN, C₂H₆ and HCOOH
 - EnhR is the slope of the linear regression when the two species are plotted against each other
 - To account for the travel time decay and to calculate the emission ratio (ER), the following equation is used:

$$ER_X = EnhR_X \left(\frac{e^{t/\tau_X}}{e^{t/\tau_{CO}}} \right)$$

ER and EF

- ER: Emission Ratio
 - Enhancement ratio with plume aging correction
 - Since travel times are accounted for, it is not location specific
- EF: Emission Factor
 - Defined by:
$$EF_X = EF_{CO} \cdot ER_X \cdot \left(\frac{MW_X}{MW_{CO}} \right)$$
 - where EF_{CO} is the emission factor of CO, which is 127 ± 45 g/kg (Akagi et al., 2011), and MW are the molecular weights
 - Not location specific

EnhR_x Plots for HCN, C₂H₆ and HCOOH



Results

Emission ratios and emission factors from this study compared against other sources (measured with ground-based FTIRs)

	This study	Other studies using ground-based FTIRs
ER _{HCN}	0.0053 ± 0.0005	0.0054 ± 0.0022 ^a
ER _{C₂H₆}	0.0150 ± 0.0014	0.0108 ± 0.0036 ^a
ER _{HCOOH}	0.0241 ± 0.0033	0.01790 ± 0.00937 ^b
EF _{HCN}	0.65 ± 0.24 g/kg	0.66 ± 0.27 g/kg ^a
EF _{C₂H₆}	2.04 ± 0.75 g/kg	1.47 ± 0.50 g/kg ^a
EF _{HCOOH}	5.03 ± 1.9 g/kg	3.15 ± 1.46 g/kg ^b

- ^a Viatte et. al., 2013
- ^b Viatte et. al., 2015

Long-term Trend Analysis

- Trend analysis:
 - Fitting a trended Fourier series:

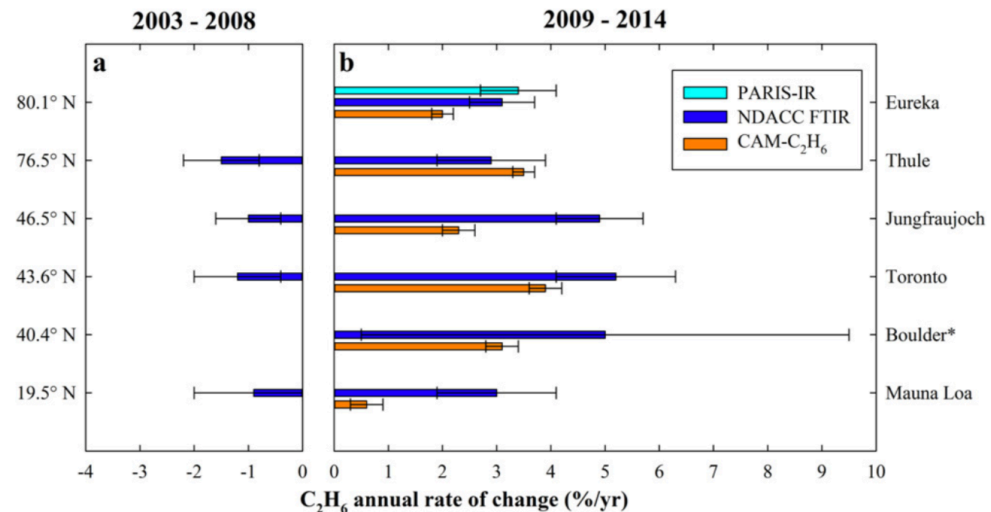
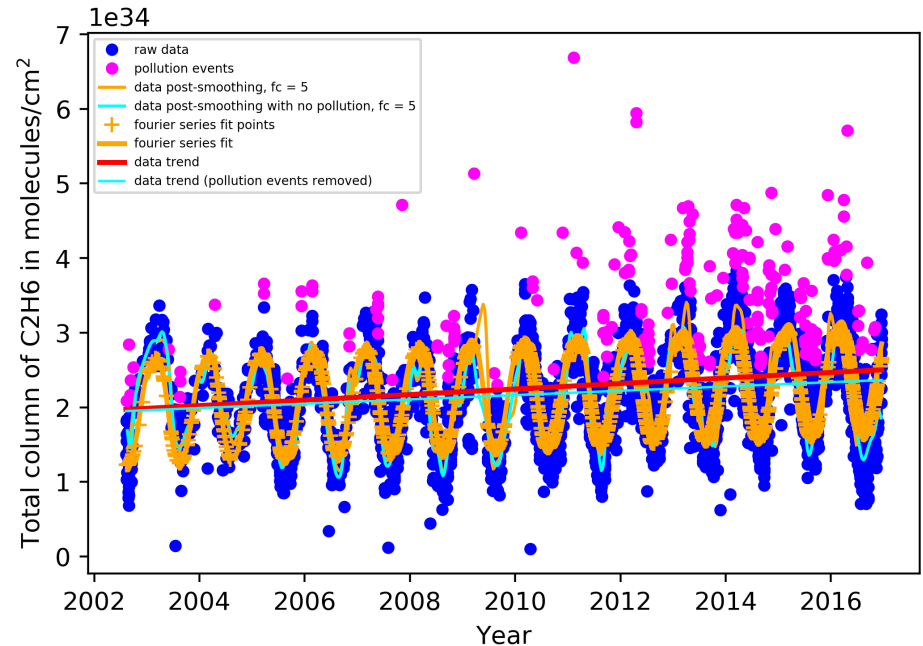
$$F(t) = a_0 + \frac{a \cdot t}{T} + \sum_{n=1}^N \alpha_n \sin\left(\frac{2\pi nt}{T}\right) + \beta_n \cos\left(\frac{2\pi nt}{T}\right)$$

- Where T is the total time of the data, and N is the order of the Fourier series (Taken to be 3 in this analysis)
- Pollution events are defined by:
 - Finding residuals, i.e., observation minus $F(t)$
 - Mirroring negative residuals as well as calculating the standard deviation of the residuals
 - Any measurements that are 2 standard deviations or higher than the fit and residuals are considered pollution events

Analysis:

Preliminary Results

- Top: C₂H₆ time-series data with pollution events and fitted trend-line
- According to this analysis:
 - Raw linear trend is: 1.72 ± 0.09 %
 - Linear trend with pollution removed is: 1.40 ± 0.23 %
 - The number of measurement years needed for trend detection (Weatherhead et al., 1998)
 - All data: 23.2 years
 - Pollution events removed: 24.4 years



Future Work

- Further analysis of biomass burning plumes over Toronto
- Trend analysis of hydrocarbons and pollutants
- Analysis of pollution events with a focus on O_3 and CO
- Integration of measurements and modeling (GEOS-Chem) to assess determinants of air quality in Toronto
 - In collaboration with Dylan Jones (UofT)
- Comparison of urban (TAO) and rural (ECCC Egbert) FTIR measurements
 - Egbert is about 90 km north of downtown Toronto (TAO)
 - Egbert FTIR is in good working condition, and currently installing a new sun tracker

Egbert FTIR & Sun Tracker



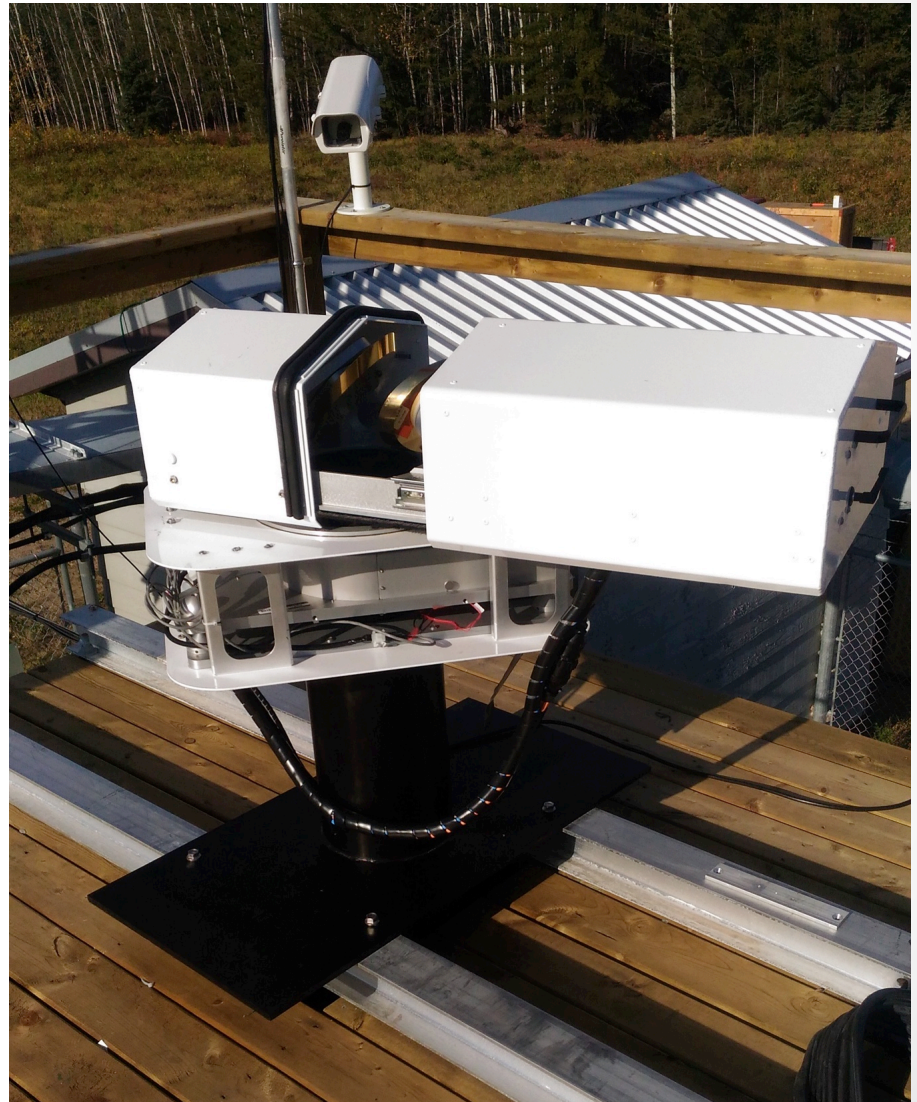
Egbert FTIR & Sun Tracker



Bruker solar tracker to be installed here

Egbert FTIR & Sun Tracker

Bruker A547 solar tracker (same model as the one to be installed at Egbert) at the East Trout Lake site



Acknowledgements

- Technologies for Exo-Planetary Science (TEPS), NSERC
- CFI, CSA, ECCC, ORD CF, CRES Tech, University of Toronto, ABB Bomem, CFCAS, PREA
- Many interns, students and postdocs who have made measurements, including : Orfeo Colebatch, Lei Liu, and Erik Lutsch

Additional Slides: Bootstrap Reanalysis

- Bootstrap resampling is done to find confidence intervals
 - It is an analysis where residuals are randomly redistributed to form a “new” set of data, where another line is refitted
 - This process is repeated (over several hundred to thousands of times)
 - The ensemble of data is analyzed to find the 2- σ confidence interval (2.5% to 97.5% coverage of the data)

(Gardiner et al., 2008)

Additional Slides: Weatherhead Method

- Detection of long-term, linear trends is affected by a number of factors
 - Size of the trend to be detected
 - Time span of the data
 - Magnitude of variability and autocorrelation
- The number of years of data needed to detect a trend strongly depends on (and increases with) the magnitude of variance and the autocorrelation coefficient
- Environmental time series data are often autocorrelated

(Weatherhead et al., 1998)

Additional Slides:

More on Biomass Burning

- “As burning occurs, it can release hundreds of years worth of stored carbon dioxide into the atmosphere in a matter of hours”
 - NASA, Earth Observatory (earthobservatory.nasa.gov)
- “Exposure to biomass burning particles is strongly associated with cardiovascular disease, respiratory illness, lung cancer, asthma and low birth weights”
 - Stanford News, 2014 (news.stanford.edu)

Additional Slides:

Atmospheric Formic Acid

- Formic acid is one of the most abundant acids in the atmosphere
- Greatly contributes to free precipitation acidity
- Affects aqueous phase chemistry
 - pH-dependent reaction rates
- Naturally produced photochemically, though emission sources (including anthropogenic) also certainly exist

(Millet et al., 2015)

Additional Slides: Retrieval

- A retrieval algorithm called SFIT4 is used to derive vertical profiles and/or columns of trace gases from their absorption and emission spectra:
 - Identify spectral lines of interest
 - Generate a model atmosphere using meteorology data from NCEP and WACCM
 - Use a forward model to simulate a model spectrum
 - Iteratively adjust the *a priori* VMR profile until the model spectrum agrees with the measured spectrum
- SFIT4 is an optimal estimation method (OEM) analysis
 - It uses both measurement data and the *a priori* information
 - It assigns weights to the *a priori* information and the measurement based on each of their uncertainties